

## Nomenclature Even answers

### MNM

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|------------------------|---------------------------|
| 2. Potassium Oxide     | 12. $\text{Li}_2\text{S}$ |
| 4. Calcium Chloride    | 14. $\text{BaF}_2$        |
| 6. Barium Fluoride     | 16. $\text{NaBr}$         |
| 8. Magnesium Iodide    | 18. $\text{Li}_2\text{O}$ |
| 10. Beryllium Selenide | 20. $\text{RbI}$          |

### TmNm

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|-------------------------|-----------------------------|
| 2. Copper (II) Chloride | 12. $\text{Cu}_3\text{N}_2$ |
| 4. Manganese Sulfide    | 14. $\text{Hg}_2\text{O}$   |
| 6. Nickel (II) Fluoride | 16. $\text{CuO}$            |
| 8. Silver Phosphide     | 18. $\text{NiBr}_2$         |
| 10. Mercury (II) Oxide  | 20. $\text{ZnO}$            |

### NmNm

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|----------------------------|--------------------|
| 2. Arsenic TriChloride     | 12. $\text{PCl}_3$ |
| 4. Diphosphorous pentoxide | 14. $\text{NO}$    |
| 6. Xenon Hexafluoride      | 16. $\text{SO}_3$  |
| 8. Nitrogen Trioxide       | 18. $\text{PCl}_5$ |
| 10. Carbon monoxide        | 20. $\text{NI}_5$  |

### Compounds with Polyatomic Ions

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|------------------------|---|
| 2. Barium Sulfate      | 12. $\text{ZnNO}_2$                               |
| 4. Chromium Phosphate  | 14. $\text{NaHCO}_3$                              |
| 6. Barium Hydroxide    | 16. $\text{CuCO}_3$                               |
| 8. Potassium Cyanide   | 18. $\text{Fe}(\text{C}_2\text{H}_5\text{O}_2)_3$ |
| 10. Calcium Dichromate | 20. $\text{Sr}_3(\text{PO}_4)_2$                  |

### Acids

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|-----------------------|---|
| 2. Sulfuric Acid      | 12. $\text{HClO}$                                 |
| 4. Phosphoric Acid    | 14. $\text{H}_2\text{CO}_3$                       |
| 6. Hydroiodic Acid    | 16. $\text{HF}$                                   |
| 8. Nitrogen Trioxide  | 18. $\text{PCl}_5$                                |
| 10. Potassium Cyanide | 20. $\text{Fe}(\text{C}_2\text{H}_5\text{O}_2)_3$ |